Group report

Solubility curve for a ternary system of liquids

➢ ***Objective/s:***

➢ ***Formula:*** *(Table)*

Formula Number	Ingredients	Quantities

➢ ***Method:***

(This section should contain any calculations related to prior experiment preparation, if none please write NA and close it)

- ***Results:***

	Volume		
	Chloroform (ml)	**Water (ml)**	**Ethanol (ml)**
1	5	0.1	0
2	5	1	
3	5	2	
4	5	3	
5	5	4	
6	5	5	
7	5	7	
8	5	10	
9	5	15	
10	5	20	
11	5	35	
12	5	50	

➢ ***Calculations:***

	Chloroform	Ethanol	Water
Molar mass (g/mol)	**119.4**	**46.1**	**18.0**
Specific gravity (g/ml)	**1.489**	**0.789**	**1.000**

- **Using the information in the table above, tabulate your calculation according to the following Tables: (V: volume, W: mass, W%: mass percentage)**

Point	Chloroform			Water			Ethanol		
	V (ml)	W (g)	W% (%)	V (ml)	W (g)	W% (%)	V (ml)	W (g)	W% (%)
1	5			0.1			0		
2	5			1					
3	5			2					
4	5			3					
5	5			4					
6	5			5					
7	5			7					
8	5			10					
9	5			15					
10	5			20					
11	5			35					
12	5			50					

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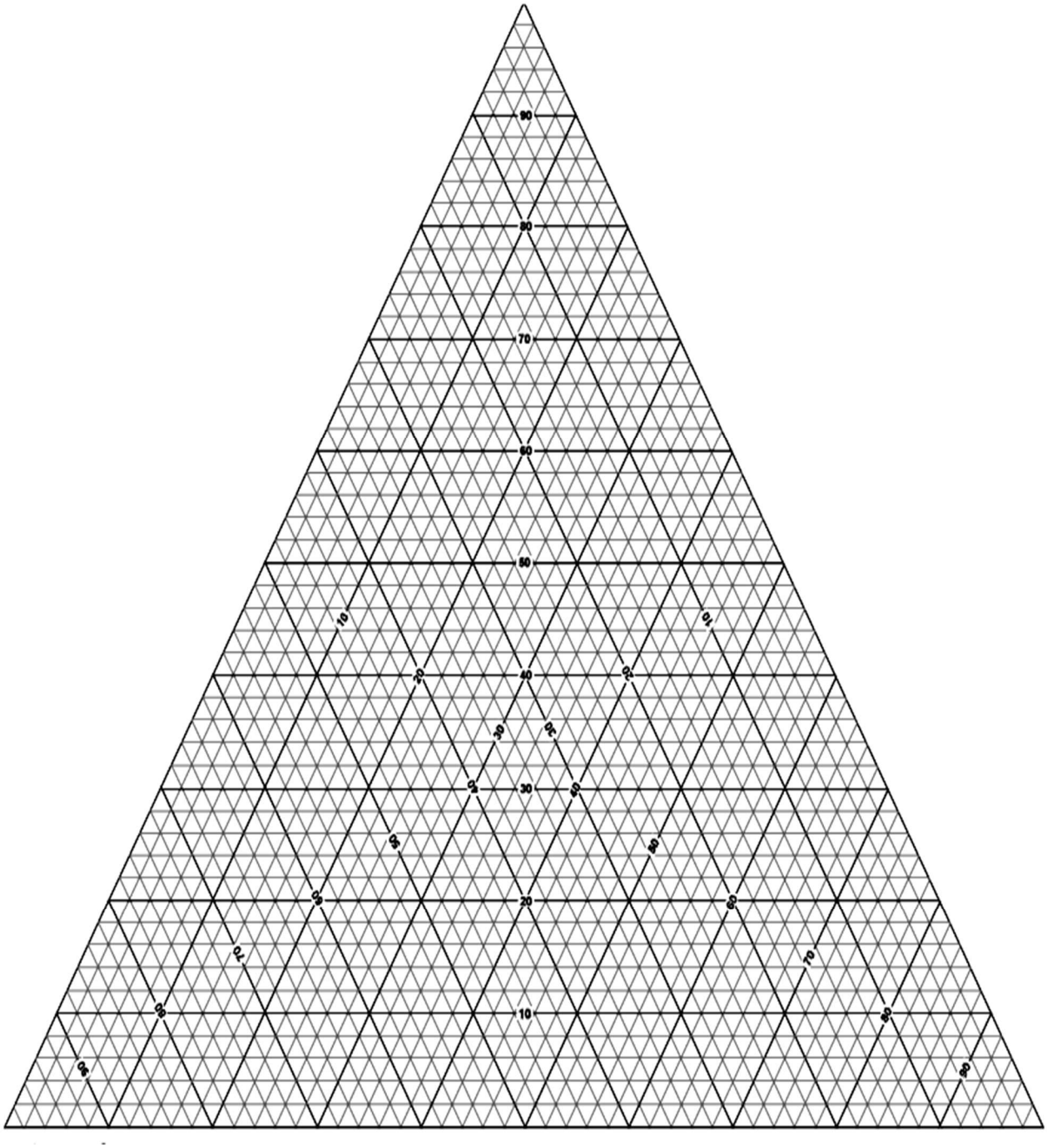
- **Draw the ternary diagram showing the binodal curve for Chloroform, water and ethanol mixtures based on mass percentage**



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- ***Discussion:***

(here add information about the following: Solubility of chloroform in water, who we determine the required volume of ethanol after each addition of water (end point determination), and describe the solubility curve of (ethanol, chloroform and water).

Conclusion:

(here conclude about: effect of ethanol on the miscibility of chloroform and water)

Experiment number: Continued from page number:

Group report

Experiment Report Workload Distribution Table

Coordinator for Current Experiment[1]:

Section[2]	Student Name[3]	Percentage of the Performed workload[4]

[1]Mention the name of the student/ group member who did arrange the work related to the current experiment group report/work management.
[2]Section or part of the group report
[3]Mention the name of student/the group member who took responsibility of the specified group report section
[4]Relative to the whole workload used to prepare the current group report.

Notes:

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Student report

Student Name: …… ……

Show your detailed calculation for **weight %** and **mole%** of chloroform, ethanol and water at one point

I choose point number :

NB.: Each student should show different example from other group members